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Self-discharge of zinc-manganese flow battery





Overview

Do aqueous zinc batteries self-discharge during idle periods?

The self-discharge of aqueous zinc batteries during idle periods remains elusive, and warranting adequate voltage and sufficient capacity is not trivial, due to the components of the battery system and the reciprocal influence among them. To investigate the origin of self-discharge, herein we construct a $Zn|$.

Are zinc-manganese dioxide batteries cathode-free?

Authors to whom correspondence should be addressed. Zinc-manganese dioxide ($Zn\text{-MnO}_2$) batteries, pivotal in primary energy storage, face challenges in rechargeability due to cathode dissolution and anode corrosion. This review summarizes cathode-free designs using pH-optimized electrolytes and modified electrodes/current collectors.

Are aqueous zinc batteries self-dischargeable?

Aqueous zinc batteries (AZBs) are a promising power storage technology for electricity storage in applications requiring high safety and power density. However, because of the use of aqueous electrolytes, AZBs face problems of self-discharge caused by a number of material and system design requirements.

What is a zinc-based flow battery?

Zinc-based flow batteries are considered to be ones of the most promising technologies for medium-scale and large-scale energy storage. In order to ensure the safe, efficient, and cost-effective battery operation, and suppress issues such as zinc dendrites, a battery management system is indispensable.



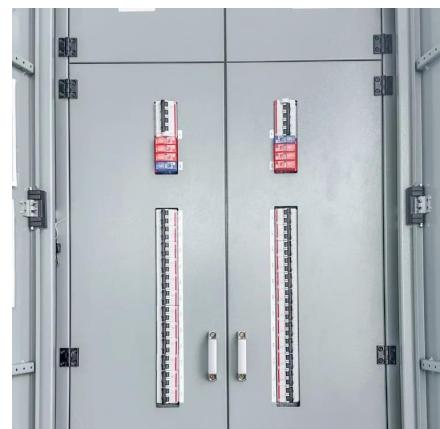
Self-discharge of zinc-manganese flow battery



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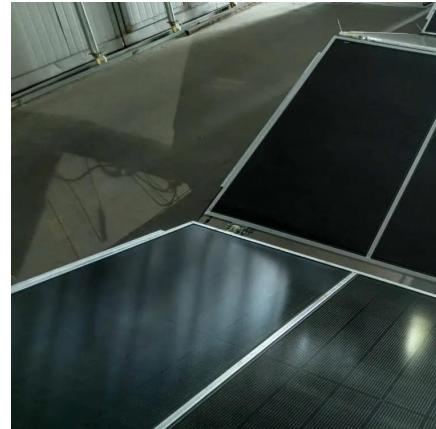
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